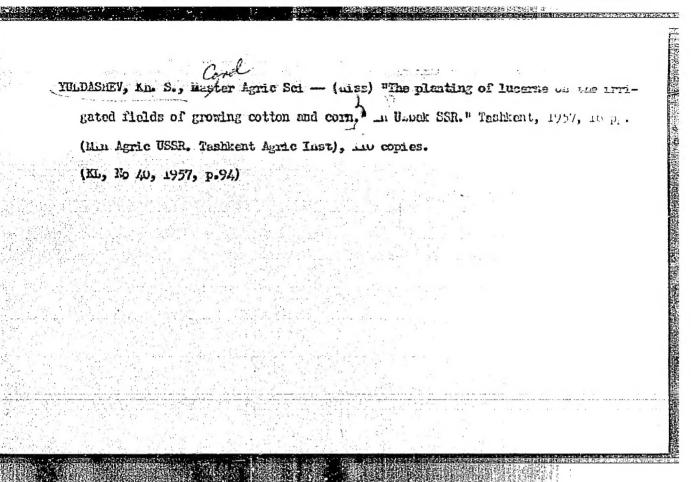
YULDASHEY, Rhalil; LANDA, L.M., kend.istor.nsuk, otv.red.; AKBAROV, A., red.; SALAKHUTDINOVA, A., tekhred.

[From the history of the development of socialist industry in Tashkent] Iz istorii razvitiis sotsialisticheskoi promyshlennosti Tashkents. Otvetstronnyi red. L.M.Ienda. Tashkent, Gos. izd-vo Uzbekskoi SSR, 1960. 175 p.

(MIRA 14:1)

(Tashkent--Industries)



YULDASHEV, K.Yu.; TSUKERVANIK, I.P.

Reactions of phenylacetylens and 2-methyl-1-phenylacetylens with anisole. Zhur. ob. khim. 34 no.8:2647-2652 Ag '64. (MIRA 17:9)

1. Tashkentskiy gosudarstvennyy universitet im. V.I. Lenina.

ATAMHANOV, E.I.; KHARAT'YAN, A.M.; BUDYANSKIY, M.V.; YULDASHEV, U.I.; SHAMSUTDINOVA, R.K.; YULDASHEV, K.Yu.

State of some metabolic indices in peptic ulcer of the storach and duodenum and the effect on them of hydrolypate therapy.

Terap.arkh. no.7:85-91 Jl '62. (MIRA 15:8)

1. Iz kafedry propedevtiki vnutrennikh bolezney (zav. - chlenkorrespondent AMN SSSR i AN Uzbekskoy SSR prof. E.I. Atakhanev) pediatricheskogo i sanitarno-gigiyenicheskogo fakul'tetev Tarikentskogo meditsinskogo instituta. (PEPTIC ULCER) (PROTEIN HYDROLYSATES) (NITROGEN METABOLISM)

KHARATIYAN, A.M.; YULDASHEV, K.Yu.

Characteristics of the amino acid composition of protein hydrolysate produced by Central Institute of the Order of Lenin of Hematology and Blood Transfusion. Probl. gemat. 1 perel, krovi 10 no.2: 52-55 F 164. (MIRA 19:1)

1. Kafedra propedevtiki vnutrennikh bolezney (zav. - chlen-korrespondent AMN SSSR 1 AN UzSSR prof. E.I. Atakhanov) sanitarn6-gigiyenicheskogo i pediatricheskogo fakul'tetov Tashkentskogo meditsinskogo instituta.

YULDASHEY, K. YU

Card) 1 1 40, 151 - 16, 1

Authors : Tsukervanik, I. F., and Yulitashev, I.

Title : Fromination of L,1-lichemylethane

Periodical : Shur. ob. khis. 24 -, 1566-1568, -

Abstract : Bromination of i,1-daymenylethane is a was carried out in a quartz flack in a bulb at 100 - 2000. Heating of the time

builb at 100 - 2000. Heating of the the ethane hydrodynous and object made. The broadmation products were a crystallization. The effect of the or

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Partition - April 17, 198.

SOY/79-29-2-15/71 Zakutskaya, M. A., Yuldashev, Kh. AUTHORS:

Condensation of o-Nitroanisole With Chloral Hydrate TITLE:

(Kondensatsiya o-nitroanizola s khloral'gidratom)

Zhurnal obshchey khimii, 1959, Vol 29, Nr 2, pp 429-431 (USSR) PERIODICAL:

The condensations of o- and p-nitrophenol (Refs 8,9) and o-nitro-ABSTRACT: anisole with chloral hydrate (Ref 10) are known among the direct syn-

theses of nitrodiaryl trichloroethanes. D. A. Shirley obtained only a yield of 12% 1,1,1-trichloro-2,2-di-(4-methoxy-3-nitro-phenyl)-ethane (II) from o-nitroanisole (0.5 mol) with chloral hydrate (0.25 mol) in the presence of concentrated sulfuric acid (50 gr) and 20% oleum (100 gr). No secondary products form-

ing in this condensation are mentioned in this connection On synthesizing trichlorinated carbinols and on investigating their reactions the authors studied this condensation more closely On varying the quantity of sulfuric acid and its rate of addition

they obtained from nitroanisole (0.2 mol), chloral hydrate (0.1 mol) and sulfuric acid (1.5 mol) the compound (II) in a yield of 60% and a small amount (5%) 1,1,1-trichloro-2,2-(4-

methoxy-3-nitrophenyl)(2'-methoxy-3'-nitrophenyl)-ethane (III) With a smaller amount of H_SO, (0.41 mol to 0.1 mol o-nitro

Card 1/2

Condensation of o-Nitroanisole With Chloral Hydrate

SOV/79-29-2-15/71

anisole) the authors succeeded inseparating 6% of the intermediate product (I) (4-methoxy-3-nitrophenyl trichloromethyl carbinol). In some condensations of o-nitroanisole (0.2 mol) with chloral hydrate (0.1 mol) and sulfuric acid (1.5 mol) only traces formed of (I), while compound (IV), melting only at 350°, formed as the chief product. Its exidation product above the positive reaction on an anthraquinone nucleus. With repair to the papers by Quelet (Ref 12) and collaborators the authors suppose the compound (IV) to be the product of autocondensation of (I) and to be a bis-(mesotrichloromethyl)-dimethoxy dimitro dihydro anthracene (Scheme). There are 17 references, 3 of which are Soviet.

ASSOCIATION:

Sredneaziatskiy gosudarstvennyy universitet ((Soviet) Central

Asian State University)

SUBMITTED:

January 8, 1958

Card 2/2

Condensation of 1	-bromo-2-methyl-1-proper	ne and 1-chloro-1-butes	ne }
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1. Tashkentskiy g 2. Chlen-korrespond (Propend	osudarstvennyy universi- ndent AN UzSSR (for TSu a) (Butens)	tet im. V.I. Lenina. kervanik). (Benzene)	
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IK, I.P., YULDASHEV, Kh. Yu.	
Condensations of vinyl halides with toluene and anis Zhur. ob. khim. 31 no.3:858-861 Mr '61.	ole. (MIRA 14:3)
1. Tashkentskiy gosudaratvennyy universitet. (Vinyl compounds) (Toluene) (Anisole)	ር <u>\$</u>

YULDASHEV, Kh.Yu.; TSUKERVANIX, I.P.

Reactions of chlorostyrene with benzene. Zuur.ob.khim. 32 no.4:
1293-1296 Ap '62. (MIRA 15:4)

1. Tashkentskiy gosudarstvennyy universitet.
(Styrene) (Benzene)

AMIHOV, Alim Muminovich, doktor ekonom.nauk; YULDASHEV, M.Tu., doktor istoricheskikh nauk, red.; AKSKL'ROD, N.B., red.; BAKHTIYAROV, M., tekhred.

[Economic development of Central Asia; from the second half of 19th century to the First World War] Ekonomicheskoe razvitie Srednei Azii; so vtoroi poloviny XIX stoletiia do pervoi mirovoi voiny. Tashkent, Gos.izd-vo UzSSR, 1959. 295 p.

(MIRA 12:8)

(Soviet Central Asia -- Economic conditions)

KARIMOV, A.K.; OSIPOVA, E.Ye.; YUIDASHEV, M.

Bitumen potential of Mesozoic sediments in the Ust-Urt.
Uzb.geol.zhur. 6 no.2:38-45 62. (MIRA 15:4)

1. Institut geologii i razrabotki neftyanykh i gazovykh mestorozhdeniy Ali Uzbekskoy SSR. (Ust-Urt-Bitumen-Geology)

YULDASHEY, P.A.; YUNUSOV, S.Yu,

Structure of vincanine. Uzb.khim.zhur. 7 no.1:44-49 '63. (MIRA 16:4)

l. Institut rastitel'nykh veshchestv AN UzSSR. (Vincamine)

Maksudo	MAKSUDOV, N.Kh.; POGORELKO, I.P.; YULDASHEV, P.Kh. Chemical investigation of Artemisia scoparia. 6 no.5:84-86 %2.			paria. Uzb.ki (1	nim.zhur. MIRA 15:12)
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YULDASHEV, P.Kn.

YULDASHEV. P. "Investigation of the Alkaloids of Vinca erecta."
Published by the Acad Sci Uzbek SSR. Acad Sci Uzbek
SSR. Inst of Chemistry. Tashkent. 1955.
(DISSERTATION FOR THE DEGREE OF CANDIDATE IN
CHEMICAL SCIENCE).

So.: Knizhnaya letopis' No. 27, July 2, 1955.

TUNUSCY, S.Yu., akademik; YULDASHEV, P.; PLEKHANOVA, N.V.

Study on alkaleids from the absveground portion of Vinca erecta
Rgt. et Schmalh. Dokl. AN Uz. SSR no.7:13-15 '56.

(HIRA 12:6)

1.Akademiya nauk UzSSR (for Yumusov).

(Alkaleids) (Vinca)

经制度和基金控制性的复数形式的特殊的现在分词 地名美国西班牙斯拉斯斯拉斯斯拉斯斯拉斯斯拉斯

In the article, "Reserpicin From Vipta erects," 5. Ye. funusor, Academician of the Academy of Sciences Uzbek SSR and P. Kh. Yuldashev of the Institute of Chemistry, Academy of Sciences Uzbek SSR, describe the method of isolation of the alkaloid reserpicin from Vinca erecta, a plant of the Apocynaceae family closely related to the plant Rauvolfia. A total of 2.6 percent of alkaloids are extracted with ether from the roots of the plant. These include the alkaloids vinkanian-3-08_20Ng, vinkadicin-2-08_00Ng, and reserpinia-208_00Ng; reserpinin is seponified with an alkali to form reserpinic acid. The acid and its nitrate are then methylated with diasomethane to obtain the pure alkaloid. (Doklady Akademii Nauk Uzbekskoy SSR, No 9, 1956, pp 23-25).

YULDASHEV
YUNUSOV, S.Yu.; Yuldashev, P.Kh.

Study of the alkaloids extracted from Vinca erecta Egl. et Schmalt.
Zhur.ob.khim. 27 no.7:2015-2018 Jl '57. (MIRA 10:10)

1.Institut khimii rastitel'nykh veshchestv i khlopka AN Uzbekskoy
SSR. (Alkaloids) (Apocynaceae)

UBAYEV, Kh.; YULDASHEV, P.Kh.; YUNUSOV, S.Yu.

Study of alkaloids of Pedicularis olgae RGL. Uzb.khim.zhur. 7 no.3:
33-36 '63. (MIRA 16:9)

LInstitut khimii rastitel'nykh veshchostv AN UzSSR.

(Figwort) (Alkaloids)

YAGUDAYEV, M.R.; RASHKES, Ya.V.; YULDASHEV, P.Kh.

Infrared spectra of vincanine and its derivatives. Uzb. khim. zhur. 7 no.6:54-58 '63. (MIRA 17:2)

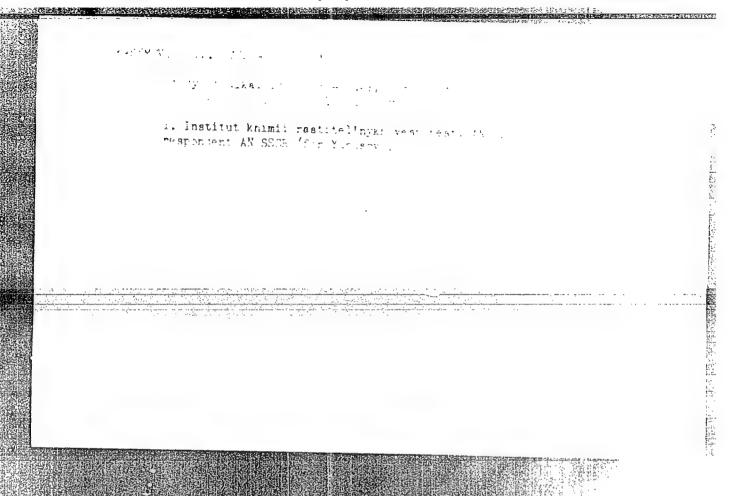
1. Institut khimii rastitel'nykh veshchestv AN UzSSR.

YULDASHEV, P.Kh.; YUNUSOV, S.Yu.

Vincarine, a new alkaloid from the roots of Vinca erecta RGL. et Schmalt.

Dokl. AN SSSR 154 no.6:1412-1413 F '64. (MIRA 17:2)

1. Institut khimii rastitelinykh veshchestv AN UZSSR. 2. Chlen-korrespondent AN SSSR (for Yunusov).



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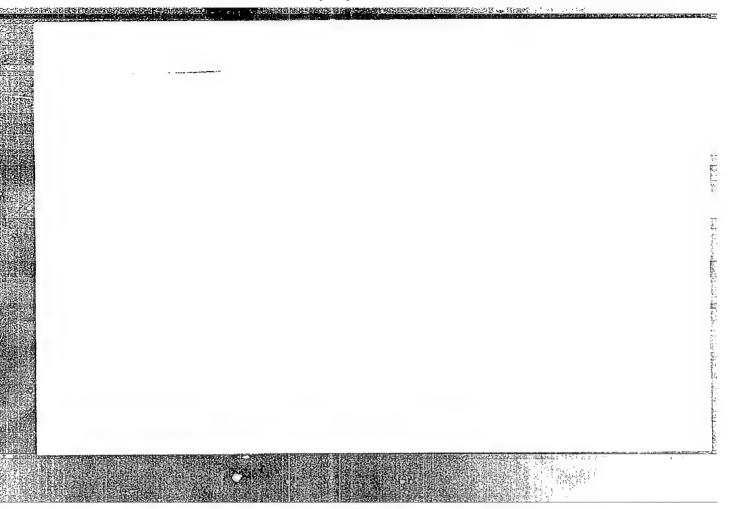
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I Instint Whimi castitelingur ver ...

KASYMOT, Sh.Z.; YULDASHET, I.Kh.; YUNUSOV, S.Yu.

Structure of vincrine and vineridine. Doll. AN SESR 1/3 no. :1240/ hg 165.

1. Institut khimii rastitelinykh veshchestv AN UzSSR. 2. Chlen-korrespondent AN SSSR (for Yunusov).



LUTFULLIN, K.L.; YHLDASHEV, P.Kh.; YUNUSOV, S.Yu. Study of the alkaloids of Pedicularis olgan. Structure of plantagonin and indicain. Khim, prirod. seed. no.5:365-366 '65. (MIRA 18:12) 1. Institut khimii rastitel'nykh veshchestv AN UzSSR. Submitted August 6, 1965.

MUCHRNKOVA, M.A.; YULDASHEV, P.Kh.; YUNUSOV, S.Yu.

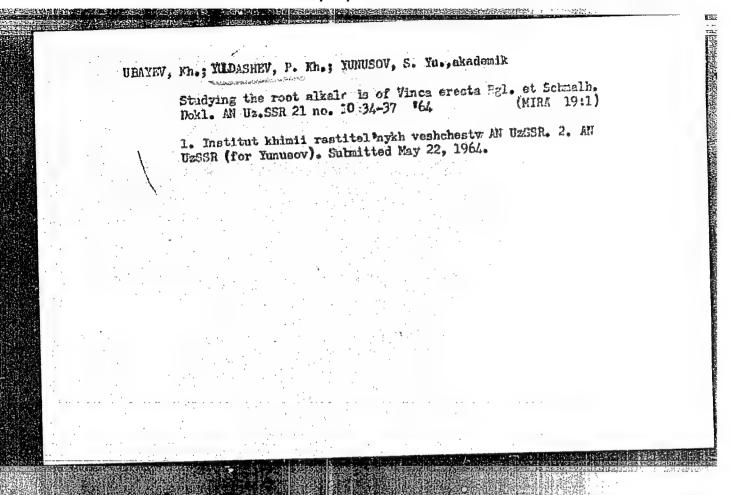
Vinervine, a new alkaloid from the above-ground part of Vinca erecta RGL et Schmalh. Izv.AN SSSR.Ser.khim. no.12:2152-2155 '65. (MIRA 18:12)

1. Institut khimii ras*itel'nykh veshchestv AN UZSSR. Submitted July 29, 1943.

YULDASHEV, P.Kh.; YUNUSOV, S.Yu.

Derivatives of vincanine. Uzb.khim.zhur. 8 no.4:61-64 164. (MIRA 18:12)

1. Institut khimii rastitel nykh veshchestv AN UZSSR. Submitted December 20, 1963.



UBAYEV, Kh.U.; YULDASHEV, P.Kh.; YUNUSOV, S.Yu.

Structure of wincanidine, alkaloid of Vinca erecta Rgl et
Schmalh roots. Izv. AN SSSR, Ser. khim. no.11:1992-1995
165.

1. Institut khimli rastitel nykh veshchestv AN UzSSR.

L 25617-66 EWT(1)/RO ACC NR: AP6016111 INTROF DEED IT ATTHOR: Ubayev, Kh. U.; Yuldashev, P. Kh.; Indexy, T. The lastitude of the Chemistry of Flant Survey , the tital night weshchesty AN UZSCR That Printers of vincanthing- as dealed to on to AN To he dispersion of entry promises The to dealaid, W operance, IR operance, or The nature of the ultraviolet and infrared are rotation, and color indicate the presence in the vincan' chromophore system of alpha-methyleneindoline, conjugate which is confirmed by the production of decomposition of beating with 20% bydrochloric seid yielded so indolenia. the indolenine base with zinc and sulfuric sold in abatation sodium borohydride in soid medium yielded a crystalline ... indoleniae base, when reduced with sodium borohydride a platinum cetalyst according to Adams in alcohol, forma a complex chromophore system is characteristic of alkalo. 'abusaicine type. Work on the establishment of the position hydroxyl is continuing. [3PRS] THE COMPANY OF A SUBMIDATE: DOWNERS IN CRITICISES OF Coro 1/2 /

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 THOR: Yulin, M. K.; Vol'epshteyn, A. P.
  : Institute of Mineral Fuels (Institut gory anily 1984,
IIIIE: Processing of liquid alkyl thenols bitained from the c
t rylpherol
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             The authors describe a processing of alkyl that I
alkylation of phonol with isobutyl alrehols, developed it am
of typermeducts. The processing is carried out in a reactor of
After driving off water, low-boiling compounds (includy) also
and phenols at a temperature up to 1950 at atmospheric process
eas manied to 2150, and the dealkylation was performed at the
Promo. The yield of anhydrous dealkylation promults was "
referenced by gas-liquid chromatography, is fully templated
and was milicone oil. P-tert-Butylphenol (PTFP) was required
fraction by crystallization and centrifuging in Public yield.
the fraction contained 35.1% PIRP, and phenol. o-temp-battles
latylphenol. The percentage compositions of the products of the
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A THORE Earymov, Gb. 2.: Yuliashev. In the Yunsey. A.
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1. (Ib) Investigation of the alkacotis in the under-corpo-
SUTROB: AN SUBE. Doklady, v. 150, no. 1, 1905, 1000
NYTIC TABLE Askaloid, solvent estraction, plant contact.
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in the Verkhne-Chirchikskiy , at of the Co. Term
other extraction of the undo secand pair it :
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sethoxyl groups. This alkaloid was new year was
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werel part gave vincanidire and acuarmica, a
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EWT (m) RM
ACC NR. AP6026897 SOURCE CODE: UR/0062/65/000/012/2152/2155
AWTHER: Kuchenkova, M. A.; Yuldashev, P. Kh.; Yunusov, S. Yu.
The tree of the them stry of Venetable Watton AN Watton
TITIE: Vinervine a new alkaloid from the above ground portion of Vinca erecta
SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 12, 1965, 2152-2155
ABSTRACT: Specimens of the plant Vinca erecta, collected in Southern Uzbekistan,
were used for cold other extraction of alkaloids, which were then divided into
phenolic and nonphenolic fractions. This resulted in the subsequent isolation
of a new phenol base, vinervine (C20H22O3N2) (m.p. 154-155°C) which is unstable to light or when in solution, and contains one OCH2 group and the substable
to light or when in solution, and contains one OCH3 group and two active H
of polarized light, which points to the base caused a pronounced levorotation
nothylenoindoline connected to a construction of alpha-
present in vinervine, one is a phenol hydroxyl and two, are esters. Heating
vinervine in ansealed and evacuated amounts with 15% HC1 for 2.5 hr at 100°C
led to the formation of a crystalline indolenine base with a melting point of
85-187°C, which proved to be identical to the indolenine base with a melting point of incanidine (another base present in V. create)
incanidine (another base present in V. erecta). The investigation of the
tructural position of the phenol hydroxyl is continuing. Orig. art. has:
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SUB CODE: 07 / SUBM DATE: 29Jul63 / ORIG REF: 002 / OTH REF: 002
UDC: 547.94
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ACC NR: AP7011362

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SOURCE CODE: UR/0393/66/000/604/0293/0294

AUTHOR: Aripov, Rh. N.; Shakirov, T. T.; Yuldashev, P. Kh.

ORG: Institute of Chemistry of Vegetable Motter, Academy of Sciences USSR (Institut khimii rastitel nykh reshchestv AN UZSSR)

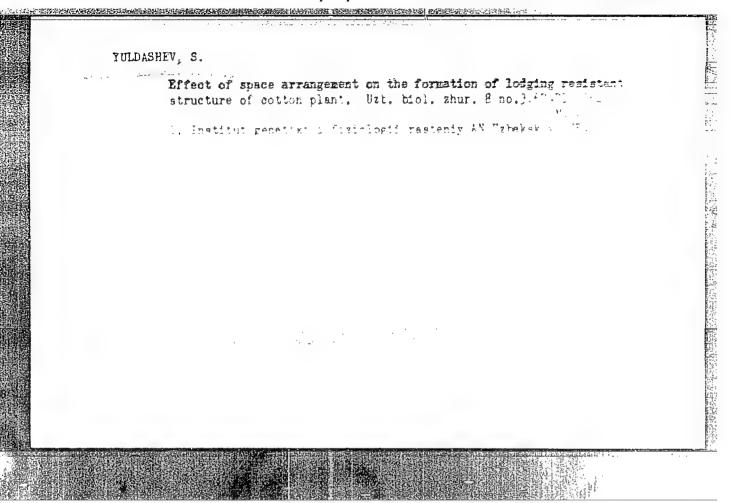
TITLE: Isolation of vincanine

SOURCE: Khimiya prirodnykh soyedineniy, no. 4, 1966, 293-294

TOPIC TAGS: plant chemistry, acetic acid

SUB CODE: 07,06

ABSTRACT: Vincanine was extracted by a countercurrent mothod from the roots of Vinca erecta Rgl. et Schmalh with a 1 percent solution of acetic acid. The extract was desorbed with 1.5 percent ammoniacal solution in 85 percent ethyl alcohol. The condensed elcoholic solution was acidified with concentrated hydrochloric acid and evaporated to remove alcohol, while the acid solution was alkalized with excess 30 percent caustic soda, and extracted three times with chloroform. The latter was distilled under vacuum to dryness and, after treatment with acetone, vincanine was isolated and converted into vincanine hydrochloride. (JPRS: 40,351)



YULDASHEY, Sh.G.; MUKHTAROY, B.H.

Case of acute psychosis caused by Taeniarhynchus infestation.

Med.zhor.Usb. no.10:84-85 0 *58. (MIRA 13:6)

1, Iz Bukharskoy oblastnoy bol'nitsy (glavnyy vrach - I.I.

Aminov). (TAPZWORMS) (MESTAL ILLESS)

YULDASHEV, S.Kh.; AKCHURINA, N.A.

Role of carbohydrates in the lodging of cotton plants. Uzb. biol. zhur. 7 no.6:67-73 '63. (MTA 17:6)

1. Institut genetiki i fiziologii rasteniy AN UzSSR.

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YULDASHAV, U, (g. Fergana).

The rank of progressive workers is increasing. Pron. koop. 12 no.3:
(6 Mr '58. (MIRA 11:3))

1. Predsedatel' pravleniya oblpromsoveta,
(Fergana Province—Gooperative societies)
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YULDASHEV, U.I.

Vitamin B-12 level of the blood serum and serum iron in anemias of gastrointestinal origin. Probl. genat. i perel. krovi 5 no.2:8-1? F '60. (MIRA 14:5)

1. Iz kafedry propedevtiki vmutrennikh bolezney (zav. - prof. E.1. Atakhanov) pediatricheskogo i sanitarno-gigiyenicheskogo fakul'tetov Tashkentskogo meditsinskogo instituta.

(CYANGCORALAMINE) (IRON IN THE BODY)

(ANEMIA)

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TULDASHr	EV, U.I., assistent	
	Concentration of vitamin B ₁₂ and iron in blood serum in diseases of the liver. Med. zhur. Uzb. no.3:10-12 Mr '60. (MIRA 15:2)	
	l. Iz kafedry propedevtiki vnutrennikh bolezney (zav prof. E.I. Atakhanov) pediatricheskogo i sanitarnogo i kulitetov Tashkentskogo gosudarstvennogo meditsinskogo i petituta	
	gosudarstvennogo meditsinskogo instituta. (CYANOCOMALAMINE) (SERUM) (IRON IN THE BODY) (LIVER_DISEASES)	
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ATAKHANOV, E.I., prof.; YULDASHEV, U.I., assistent

Amount of vitamin B₁₂ and iron in blood serum in storagh diseases. Mod. zhur, Uzb. no.4:8-13 Ap '61. (MIRA 14:5)

l. Iz kafedry propedevtiki vnutrennikh bolezney pediatricheskogo i sanitarno-gigiyenicheskogo fakul'tetov Tashkentskogo gosudarstvennogo meditsinskogo instituta. (STOMACH__DISEASES) (SERUM) (IRON IN THE BODY) (CYANOCOBALAMINE)

 ATARHANOV, E.I.; KHARAT'YAN, A.M.; HUDYANSKIY, M.V.; YULDASHEV, U.I.; SHAHSUTDIROVA, R.K.; YULDASHEV, K.Yu.

State of some metabolic indices in peptic ulcer of the stomach and duodenum and the effect on them of hydrolysate therapy.

Terap.arkh. no.7:85-91 Jl '62. (MIRA 15:8)

1. Iz kafedry propedevtiki vnutrennikh bolezney (zav. - chlenkorrespondent AMN SSSR i AN Uzbekskoy SSR prof. E.I. Atakhanov) pediatricheskogo i sanitarno-gigiyenicheskogo fakul'tetov Tashkentskogo meditsinskogo instituta. (PEPTIC ULCER) (PROTEIN HYDROLYSATES) (NITROGEN METABOLISH)

YULDASHEV, U. I.

Dynamics of the vitamin B_{12} and iron content in the blood serm in the pellagra syndrome. Terap. arkh. 34 no.5:71-76 '62. (MIRA 15:6)

1. Iz kafedry propedevtiki vnutrennikh bolezney (zav. - chlenkorrespondent AMN SSSR i AN UzSSR prof. E. I. Atakhanov) pediatricheskogo i sanitarnogo fakulitetov Tashkentskogo mediteinskogo instituta.

(PELLAGRA) (CYANOCOBALAMINE) (IRON IN THE BCDY)

YULDASHEV, U.I., kend.med.nank

Content of vitamin B₁₂ and iron in the blood serum in anemia. Terap. arkh. 34 bo.12:69-74 D'62. (MIRA 16:6)

l. Iz kafedry propedevtiki vnutrennikh bolezney (zav. - chlen-'orrespondent AME SSSR i AN UZSSR prof. E.I.Atakhanov) pedistricheskogo i sanitarno-gigiyenicheskogo fakul'tetov Tashkentskogo meditsinskogo instituta. (CYANOCOBALAMINE) (IRON IN THE BODY)

YULDASHEV, Zh.

Characteristics of the petrographic composition of Mesozcic sediments in the northern Ustyurt. Uzb.geol.zhur. 8 no.3:34-38 '64. (MTRA 18:12)

1. Institut geologii i razrabotki neftyanykh i gazovykh mestorozhdeniy Gosudarstvennogo geologicheskogo komiteta SSSR. Submitted Jan. 31, 1964.

发展的技术是指在现在的是是18日的目的的特殊是并分别的原因的。

1 3 Sept.

YULDASHEV, Th.

Gross section types of Mesozote sediments in northern Usiyurt. The gool, shur, 8 no.5878-80 164. (MIRA 18:5)

1. Institut geologii i razrabotki neftyanykh i gazovykh mestorozhdeniy Gesudarstvennege geologicheskogo komitata GSSR.

CIA-RDP86-00513R001963120011-7

ACCESSION NR: AT4042432

S/3103/64/000/002/0175/0182

AUTHOR: Usmanov, Kh. U., Tillayev, R. S., Musayev, U. N., Yuldasheva, Kh.

TITLE: Thermomechanical properties and plasticizing of grafted copolymers obtained by radiation polymerization

SOURCE: AN UzSSR. Institut khimii polimerov. Khimiya i fiziko-khimiya prirodny*kh i sinteticheskikh polimerov, no. 2, 1964, 175-182

TOPIC TAGS: grafted copolymer, acrylonitrile, polystyrene, polyvinylchloride, vinyl perchloride, glass temperature, Gamma-irradiation, plasticizer, saponified copolymer, radiation polymerization, polymer plasticizing, polymer thermomechanical property

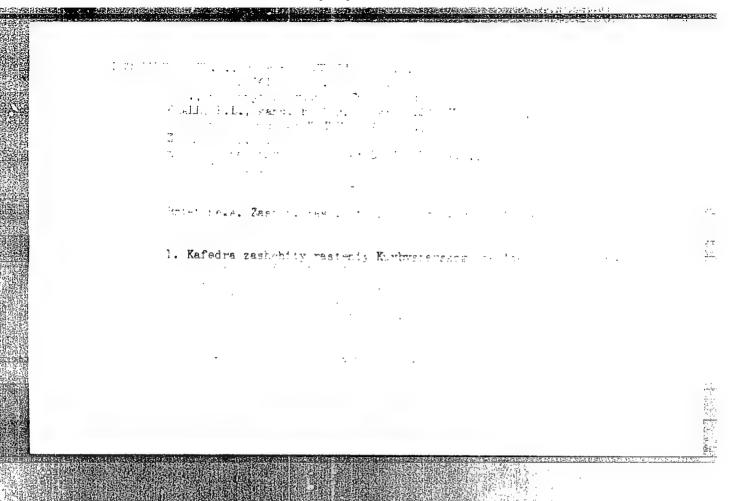
ABSTRACT: A study of the thermomechanical properties of grafted copolymers obtained by grafting acrylonitrile on polystyrene, polyvinvyl chloride and vinyl perchloride showed that the glass temperature Tc of these copolymers, regardless of the ratio of the components, corresponds essentially to the glass temperature of the initial polymers, but that the flow temperature Tt lies above the temperature of chemical stability of the products. Copolymers, as compressed tablets (3-4 mm thick and 7 mm in diameter), were tested before and after irradiation at doses of 1-10 Mr. The thermomechanical curves were plotted with the dynamometric scales of Kargin and Sogolova at a constant load for 10 sec., at a specific Card

ACCESSION NR: AT4042432

load of 1.4 kg/cm². The curves obtained for all the copolymers, with or without plasticizers were quite similar, and showed less effect of temperature than on pure polymers. Tabulated irradiation data showed that the thermomechanical properties of grafted copolymers remain almost unchanged under the influence of irradiation. This indicates the greater stability of grafted copolymers to Y-rays as well as to high temperatures. The flow of grafted copolymers is therefore considered to be almost independent of grafting. An investigation of the plasticizing of grafted copolymers showed that grafted copolymers synthesized from two homopolymers which have a common plasticizer remain unchanged in their compatibility with this plasticizer. For grafted copolymers containing, on the one hand, chains able to plasticize (polystyrene, polyvinyl chloride) and, in the other component, unplasticizable rigid chains (polyacrylonitrile), the compatibility with the plasticizer is low and limited. The change in thermomechanical properties (decrease in Tc) with increasing plasticizer concentration (tetralin or methylbenzoic ether) is plotted. In addition, analytical data for nitrogen content and acid number of the grafted copolymers are tabulated. The thermomechanical curves of saponified vinyl perchloride and polyacrylonitrile grafted copolymers showed that the glass temperature is decreased and the plasticity is increased by saponification. A further increase in plasticity is produced by plasticizers, especially glycerol. Such an increase could never be obtained by plasticizing unsaponified grafted copolymers. Orig. art. has: 2 tables and 3 figures.

2/3

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YULPASHEVA, L.K.

Category: USSR / Physical Chemistry - Electrochemistry

B-12

Abs Jour: Referat Zhur-Khimiya, No 9, 1957, 30129

Author : Fayzullin F. F., Yuldasheva L. K.

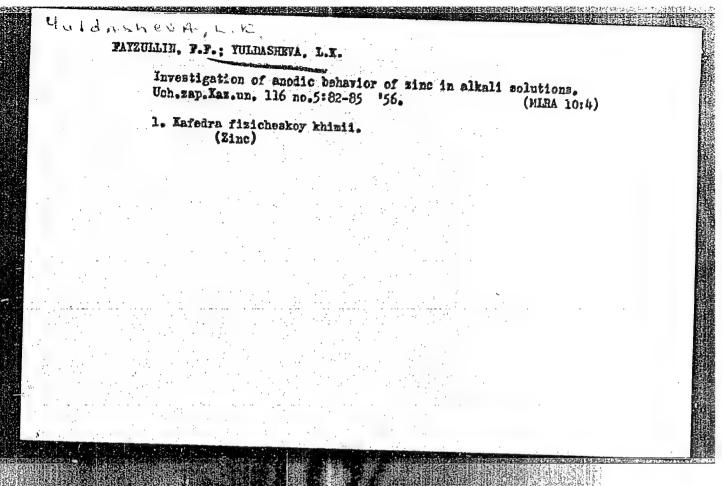
: Kazan' University

Title : Study of Ancdic Behavior of Zinc in Alkaline Solutions

Orig Pub: Uch. zap. Kazanskogo un-ta, 1956, 116, No 5, 82-85

Abstract: By the method of automatic recording of the (co, t) curves (RZhKhim, 1957, 12280) a study was made of anodic polarization of Zn in 0.25, 0.5 and 1 N NaOH at 40 and 60° and $1 = 6 \text{ a/dm}^2$. Cn application of the current the portential of Zn rises sharply and evolution of 0, begins. Cxidation is attended by periodical, very rapid, potential changes, caused by periodical breakdown and formation of oxide film. By the gravinetric method a determination was male of the rate of formation of oxide film on Zn at 1 of 6 and 12 a/dm2; an increase of i increases rate of formation of the film. On increase of the temperature there takes place a decrease in overvoltage of O, evolution, which results in an increased rate of formation of the oxidic film.

Card : 1/1



8/048/63/027/001/030/043 B125/B102

AUTHORS:

PERIODICAL:

Arbuzov, B. A., Samitov, Yu. Yu., and Yuldasheva, L. K.

TITLE:

Spectra of proton magnetic resonance of the substituted dislo-

cation

Akademiya nauk SSSR. Izvestiya. Seriya fizicheskaya, v. 27, no. 1, 1963, 89 - 92

TEXT: A study of the p.m.r. spectra of 2-methyldioxolane, 2-chloromethyldioxolane, and trichloromethyldioxolane proved that the influence of the halide replacing the hydrogen in the methyl radical of 2-methyl-1,3-dioxolane extends as far as the protons of the methylene groups that are in 0-position with respect to the oxygen. The polar groups also cause chemical shifts of the β-hydrogens. Owing to the effect of the five-membered rings the chemical shifts of the protons in dioxalane are by 0.3 smaller than in the compounds with open chains (e.g. actal, orthoester). A substitution of the proton of the methyl radical by the first chlorine atom influences the chemical shift of the protons of the methylene group more strongly than the subsequent introduction of further chlorine atoms. There is 1

Spectra of	f proton	y (S/048/63/027/ B125/B102	001/030/043	
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ASSOCIATIO	Ni Kazanskiy Lenina (Ka	gosudarstvanyy zan' State Univ	universitet im. ersity imani V. I	V. I. Ul'yanova- Ul'yanov-Lenin)	
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_ Card 2/2	, t	Sa b			

KATAYEVA, L. M.; ANGNIMOVA, I. V.; YULDASHEVA, L. K.; KATAYEV. Ye. G.

Reaction of selenols with scetylene derivatives. Part 2:

Structure of the products of interaction between selections and phenylacetylere and Ametryle-feature interaction. By no. 1996-5-1.

1. Kazanskiy gosudarstvennyy universitet imeni V. I. Ul'yanova-lenina.

(Selenophene) (Acetylene)

ARBUZOV, D.A.; YULDASHEVA, L.K. Dipole moments and the conformation of cyclic empounds. Report E.1: 1,3-Diexolanes. Izv. AN SSSR.Otd.khim.nauk no.10:1728-1754 11. ANIMATEDIA (MIRATEDIA) 1. Khimichekiy institut im. A.M.Butlerova Kazanskogo gosudarstvenn pouniversiteta. (Dioxolane-Dipole moments)

ARBUZOV, B.A.; YULDASHEVA, L.K.

Dipole moments and the conformation of cyclic compounds. Izv. AS SSSE, Otd.khim.nauk no.10:1734-1737 0 '62. (MIRA 15:10)

1. Khimicheskiy institut im. A.M. Butlerova Kazanskogo gosudarstvennogo universiteta.

(Dioxane-Dipole moments)

ARBUZOV, B. A.; SAMITOV, Yu., Yu.; YULDASHEVA, L. K.

Proton magnetic resonance spectra of substituted dioxolenes. Izv. AN SSSR. Ser. fix, 27 no.1189-92 Ja '63. (MIRA 16:1)

1. Kazanskiy gosudarstvennyy universitet im. V. I. Uliyanova-Lenina.

(Nuclear magnetic resonance and relaxation)
(Dioxolanes—Spectra)

ARBUZOV, B.A., akademik; VILICHINSKAYA, A.R.; SAMITOV, Yu.Yu.; YULDASHEVA, L.K.

Structure of allocatene dioxide. Dokl. AN SSSR 164 no.5:1041(MIRA 18:10)

1. Nauchno-issledovatel skiy khimicheskiv institut im. A.M. Butlerova pri Kazanskom gosudarstvennom universitete.

KHABIBULLIN, Sh.T.; YULDASHEVA, L.L.						
	Analysis of star counts in the dark neonlas using K.E. Ogorodnikov's method. Uch.zap.Kaz.un. 116 no.1:89-92 (HLRA 10:5)					
	1.Kafedra astronomii. (Nablulae) (Ogorodnikov, K.E.)					
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YULDASHEVA, M.

Selecting the design form of the systems of equations for unit voltages in electric systems. Izv. AN Uz. SSR. Ser. tekh. nauk 9 no.5:23-25 165. (MIFA 18:10)

1. Ngbekskiy nauchno-issledovatel'skiy institut energetiki i aytomatiki.

32598. YULDASHEVA, O. Kak moye zveno dobilos' urozhaya sakharnoy svekly po 831 tsentinery s gektara. (kog'khoz kzyl kuyash. kok-andsk, rayon fergan. obl. sots. sel. khoz-vo uzbekictana, 1949, No 3, s. 56-60

SO: Letopis' Zhurnal' nykh Statey, Vol. 44

YULDASHEVA, S.P. Study of diffusion of endamic golter in the western districts of the Ohm Valley, Izv. AN Rir. SSR, biol. nauk 2 no.6:73-77 160. (CHU VALLEX...GOITER) (CHU VALLEX...GOITER)

YULDASHEVA, S.N., kandidat meditsinskikh nauk

Sanatorium services for rhousatic children in Tashkent. Pediatriia
39 no.2:43-47 Mr-Ap '56. (MLRA 9:3)

1. Is Kliniki gospital'noy pediatrii (dir. sasluzhennyy deyatel'
nauki prof. R.S. Gershenovich) Tashkentskogo meditsinskogo instituta imeni V.M. Molotova

(RHEDMATISM, in infant and child,
management in sanatoria (Rus))

Distribution and clinical forms of rhaumatic fever in children in Uzbekiztan. Fediatrila 35 no.12:18-20 D '57. (MIRA 11:2) 1. Is kliniki gospital' noy pediatril Tashkentskogo meditsinskogo instituta (zav. - zasluzhennyy dayatel' nauki prof. R.S.Gershenovich) (UZBEKISTAN--PHEUMATIC FEVER)

LYUBETSKAYA, M.Z.; YUIDASHEVA, S.H.; NURIDDINOV, M.R.

Conditioned reflex changes in the pupil in rheumatic fever in children. Pediatriia 36 no.2:89 F 159. (MIRA 12:4)

1. Iz kliniki gospital'noy padiatrii Tashkentakogo meditsinakego instituta.

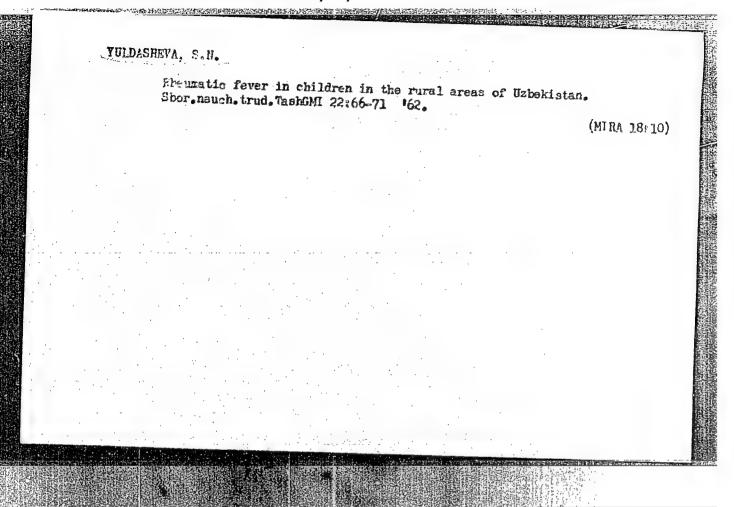
(PUPIL (ETE)) (RHEUMATIC FEVER)

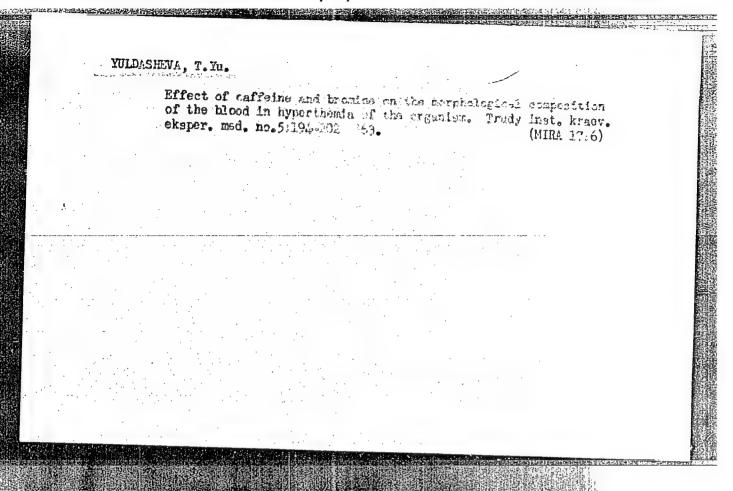
YULDASHEVA, S.N., dotsent

Decrease of rheumatic fever in children in seme cities and rural localities of Uzbekistan. Trudy Tadzh, med. inst. 50:84-86 161. (MIR4 17:8)

1. Zaveduyushcinya kafedroy fakul tetakoy pediatrii Tushkentakop meditsinskogo instituta.

Distribution of rheumatism among children in rural areas of Uzbekistan. Vop. okh. mat. 1 det. 7 no.5:57-59 My '62. (MIRA 15:6) 1. Iz gospital'nov pediatricheskov kliniki Tashkentskogo meditsinskogo instituta (zav. - prof. R.S. Gershenovich [deceased]). (UZHEKISTAN-RHEUMATIC FEVER)





Fifect of overheating and insolation on the blood picture and reticulocyte content in the blood of healthy and decorticated dogs. Uzb. biol. zhur. 7 no.5:53-55 '63. (MIRA 18:11) 1. Tashkentskiy gosudarstvennyy universitet imeni Lenina.

S/078/61/006/004/003/018 B121/B216

AUTHORS:

Sokolova, N. D., Skuratov, S. M., Shemonayeva, A. M.

Yuldasheva, V. M.

TITLE:

Determination of the standard enthalpy of formation of the

alpha and beta modification of metaboric acid

PERIODICAL:

Zhurnal neorganicheskoy khimii, v. 6, no. 4, 1961, 774-776

TEXT: The standard enthalpies of formation of the alpha and beta modifications of metaboric acid were obtained by determining the standard enthalpies of solution at 295° K. α -HBO $_2$ was prepared by heating analytical grade H $_3$ BO $_3$ for several days in an ampulla under a vacuum of 10-20 mm Hg at 90° C. β -HBO $_2$ was obtained by heating boric acid in an open ampulla to 160° C in the course of 8 hr and keeping it at this temperature for several days. X-Ray analytical data indicated the products to be the pure α - and β modifications. X-Ray analysis was made by A. A. Babad-Zakhryapin at the Institut fizicheskoy khimii Akademii nauk SSSR (Institute of Physical

Card 1/3

S/078/61/006/004/003/018 B121/B216

Determination of the standard ...

Chemistry, Academy of Sciences USSR). The measurements were carried out in a calorimeter with an adiabatic jacket. Metaboric acid was introduced into the calorimeter in closed ampullas which were then broken. The thermometer readings were correct to \pm 0.0005°. The water equivalent of the calorimeter was determined by electrical heating (\sim 171 cal/deg). The temperature rise was 0.03-0.06°C for a-HBO₂, and 0.17°C for β -HBO₂. The enthalpy of solution of a-HBO₂ was measured to be 700 and 400 mole H₂O for a final concentration of 1 mole H₃BO₃, both values agreeing within the measuring error. For β -HBO₂, the enthalpy of solution was measured at a final concentration of 1 mole H₃BO₃ to 500 mole H₂O. The enthalpies of formation of the alpha and beta modifications of metaboric acid determined at final concentrations of 1 mole H₃BO₃ to 500 mole H₂O

are α -HBO₂ Δ H₂₉₃ = + 0.47 \pm 0.01 kcal/mole Δ H₂₉₃ = + 1.76 \pm 0.01 kcal/mole

The standard enthalpies of formation of the alpha and beta modifications

Card 2/3

Determination of the standard ...

S/078/61/006/004/003/018 B121/B216

of metaboric acid from crystalline boron and gaseous oxygen and hydrogen were calculated at $\alpha\text{-HBO}_2$ $\Delta\text{H}_{\text{formation}}^0 = -189.0 \pm 0.4 \text{ kcal/mole}$

 β -HBO₂ Δ H^O_{formation} = -190.3 \pm 0.4 kcal/mole

There are 2 tables and 10 references: 3 Soviet-bloc and 7 non-Soviet-bloc.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova,

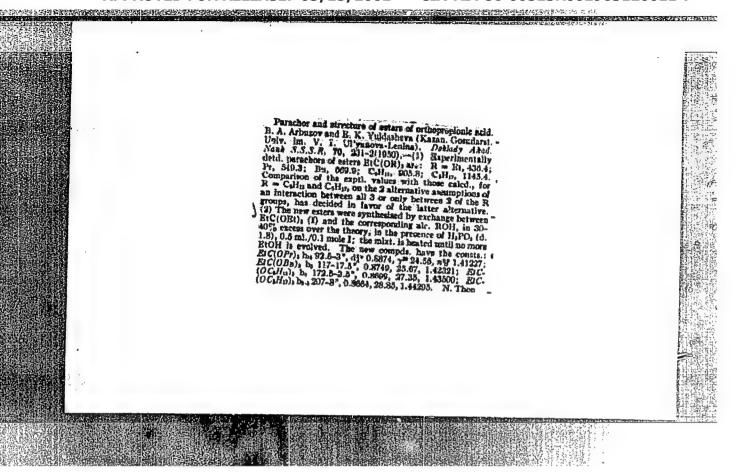
Khimicheskiy fakultet (Moscow State University imeni

M. V. Lomonosov, Chemical Division)

SUBMITTED:

March 4, 1960

Card 3/3



YUIDYBAYEV, Kh.S., veterinarnyy vrach. Use of Academician M.P. Tushnov's histolysates in vaterinary practice. Veterinariia 31 no.2:39-40 F '54. (MLRA 7:2) 1. Bashkirskaya veterinarnaya opytnaya stantsiya.
(Veterinary medicine) (Tissue extracts)

CIA-RDP86-00513R001963120011-7"

APPROVED FOR RELEASE: 03/15/2001

YULENBER, G. YE.

AUTHOR:

YULENBEK, G. Ye.

53-3-5/10

TITLE: PERIODICAL:

"In Memory of Professor P. KHRMIFEST", Russian, Uspekhi Fiz. Nauk, 1957, Vol 62, Nr 3, pp 367-370 (U.S.S.R.)

ABSTRACT:

On the occasion of being awarded the OERSTED medal, G.E. WULFABEK delivered a speech before the American Union of Teachers of

Physics in which he spoke about the great pedagogue and physicist

EHRENFEST whose pupil he had been and to whom he owes his

pedagogical successes.

ASSOCIATION:

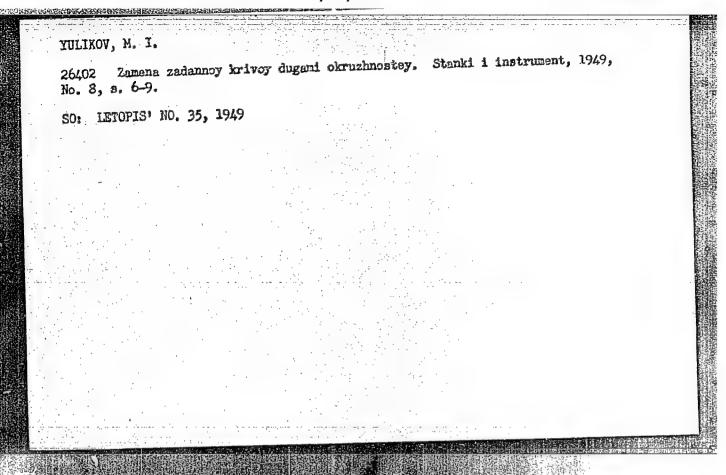
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PRESENTED BY:

SUBMITTED AVAILABLE:

Library of Congress

Card 1/1



YULIKOV, N.I.

32531. Netody pravki shlifoval'nykh krugov. stanki k kastrumont, 1909, No 10, 7, 8-11.

So: Letopis' Zhurnal' nykh Statey, Vol. 44, Moskva, 1949

XUCCHROSCOVICYOUX

"Investigation of Rams for Machining Curvilinear Profiles." Thesis for degree of Cand. Technical Sci. Sub 4 Jun 50, Moscow Order of Labor Red Banner Higher Technical School imeni N. E. Bauman

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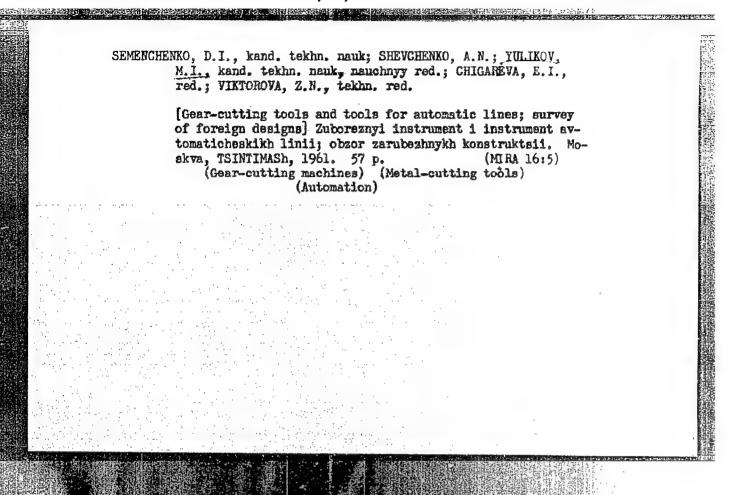
for Degrees in Science ond Engineering in Moscow in 1950. From Vechernvava Moskva, Jan-Dec 1950

TEMCHIE, Crigoriy Il'ich; LUR'IE, C.B., prof., retsenzent; YULIKOV, M.I., kand.tekhn.nauk, red.; MORCZOTA, M.B., red.; zdatel'stva; MATVEYEVA, Ye.B., tekhn.red.; KL'XIMD, V.D., tekhn.red.

[Theory and computation for satting up multiple-tool equipment]
Teoria i raschet mnogoinstrumentnykh naladak. Moskva, Gos. nauchno-tekhn.izd-vo mashinostroit.lit-ry, 1957. 555 p.

(Machine tools)

(Machine tools)



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TEMCHIN, G.I.[doceased]; YULIKOV, M.I., kand. tekhn. nauk, red.; retsenzent; ESTERZON, M.A., kand. tekhn. nauk, red.; SEMENGHENKO, V.A., red.izd-va; MODEL', B.I., tekhn. red.; DEMKINA, N.F., tekhn. red.

[Multitool adjustments; theory and design] Mnogoinstrumentnye naladki; teorila i raschet. Izd.2., ispr. Moskva, Mashgiz, 1963. 542 p. (MIRA 16:12) (Metal cutting)

ARSHINOV, V.A., kand. tekhm. nauk; ALEKSEYEV, G.A., inzh.; YEGOROV, S.V., kand. tekhm. nauk, dots., retsenzent; MALLNOVSKIY, V.R., inzh., retsenzent; TULIKOV, M.I., kand. tekhn.nauk, red.

[Metal cutting and metal-cutting tools] Rezanie metallov i rezhushchii instrument. Moskva, Izd-vo "Mashinostroenie," (MIRA 17:7)

YULIKOV, S.A.

Dolichocolon in the clinic for internal diseases. Zdravookhranenie 5 no.4:43-48 Jl-Ag '62. (MIRA 15:9)

1. Iz bol'nitsy Lechebno-sanitarnogo upravleniya Moldavskoy SSR (nachal'nik - kand.med.nauk M.G.Zagarskikh).
(COLON-ABNORMITIES AND DEFORMITIES) (MEDICINE, INTERNAL)

KHOXHLOV, A.S.; SILAYEV, A.B.; STEPANOV, V.M.; YULIKOVA, Yo.P.; TROSHKO, Ye.V.; LEVIN, Ye.D.; MAMIOFE, S.M.; SINITSYNA, Z.T.; CHI CHAN-TSIN [Ch'in Ch'ang-Ch'ing]; SOLOV'YEVA, H.K.; IL'INSKAYA, S.A.; ROSSOVSKAYA, 7.S.; DMITRIYEVA, V.S.; SEMENOV, S.M.; VEYS, R.A.; BEREZINA, Ye.K.; RUBTSOVA, L.K.

A new type of polymyxin, polymyxin M. Antibiotiki 5 no.1:3-9 Ja-F '60. (MIRA 13:7)

1. Vsesovuznyy nauchno-issledovatel skiy institut antibiotikov i laboratoriya khimii belka i antibiotikov khimicheskogo fakul'teta Moskov kogo ordena Lenina gosudarstvennogo universiteta imeni M.V. Lomonosova.

(POLYMIXIN)

YULIKOVA, WE. P., KUZMHA, N. A., SILAYEV, A. B., KATRUKHA, G. S. (USGR)

"Mechanism of Polymixin M Inactivation."

Report presented at the 5th International Biochemistry Congress, Moscow, 10-16 August 1961

SILAYEV, A.B.; STEPANOV, V.M.; YULIKOVA, Ye.P.; TROSHKO, Ye.V.; LEVIN, Ye.D. Chemistry of polymyxin M. Part 1: Qualitative amino acid analysis and analysis for end groups. Zhur. ob. khim. 31 no.1:29 -305 Ja (M. RA 17:1) . 61. 1. Moskovskiy gosudarstvennyy universitet. (Polymyxin)

SILAYEV, A.B.; STEPANOW, V.M.; YULIKOVA, Yo.P.; MPRATOVA, G.L. Chemistry of polymixin M. Part 2: Quantitative enino acid composition. Zhur, ob. khim. 31 no.3:1023-1026 Hr 61. (MIRA 14:3) 1. Moskovskiy gosudarstvennyy universitet.
(Polymixin)

> CIA-RDP86-00513R001963120011-7" APPROVED FOR RELEASE: 03/15/2001

Chemistry of polymykin M. Part 3: Partial hydrolysis of polymykin M. Zhur.ob.khim. 31 no.2:2712-2716 Ag '61. (MIRA 14:8) 1. Moskovskiy gosudarstvennyy universitet imeni M.V. Lomonosova.			
	ROMOHOSOVA.	(Polymyxin)	

SILAYEV, A.B.; STEPANOV, V.N.; YULIKOVA, Ye.P.; MICHAYLOVA, I.Yu.; (Bolgariya); UDALOVA, T.P.

Study of the inactivation of polymyxin. M. Antibiotiki 7 no.7: 638-643 J1:62. (MIRA 16:10)

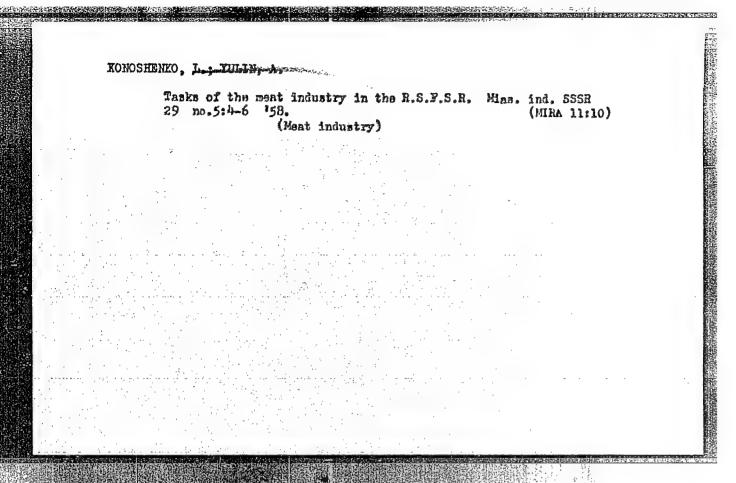
1. Laboratoriya khimii belka i antibiotikov khimicheskogo fakul'teta Moskovskogo universiteta imeni M.V. Lomonosova.

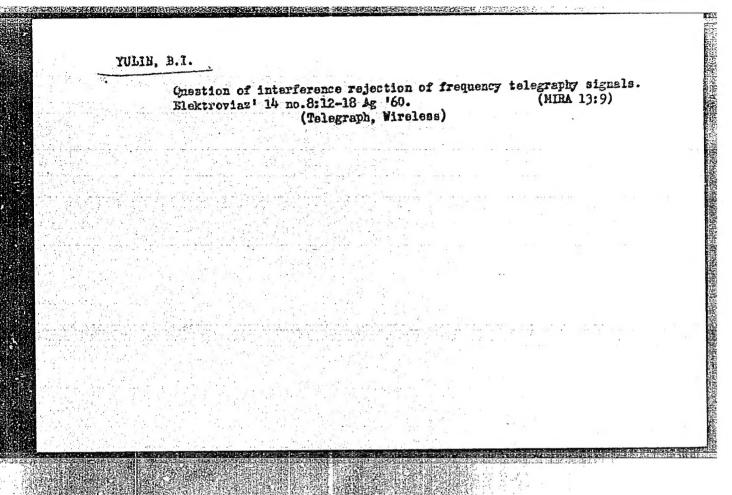
SILAYEV, A.B.; YULIKOVA, Ye.P.; BARATOVA, L.A.

Chemistry of polymyrin M. Part 5: Identification of fatty acid.

Zhur.ob.khim. 32 no.3:818-820 Mr '62. (MIRA 15:3)

1. Moskovskiy gosudarstvennyy universitet imeni M.V.Lomonosova. (Polymyxins) (Acids, Fatty)





VOL'-EPSHTKYN, A. B.; GRIGOR'YEV, S. M.; KRICHKO, A. A.; KONYASHINA, R. A.; SUROVISEVA, V. V.; YULIN, M. K.

Production of aromatic hydrocarbons from pyrolysis tar of hydrocarbon gases by hydrogenation. Trudy IGI 17:269-277 '62.

(MIRA 15:10)

(Hydrocarbons) (Coal-tar products)
(Hydrogenation)

YULIN, M.K.; VOL'-EPSHTEYN, A.B.; DAVTYAN, N.A.; LISYUTKINA, L.N.

Investigating the composition of the products of the alkylation of phenol with isobutyl alcohol and isobutene. Neftekhimia 4 no.5:717-721 S-0 *64. (MIRA 18:1)

1. Institut goryuchikh iskopayemykh AN SSSR.

L 53738-65 EPF(c)/EWT(m) Pr-4 BM ACCESSION NR: AP5015486

AUTHOR: Mekarova, T. F.; Mosnkov, P. F.; Sheshin, M. Yulin, M. K.

TIPLE: A method for the preparation of p-tert-butylish

SOURCE: Fyulleten' izobreteniy i tovarnykh znakov,

TOPIC TAGS: tert butylphenol synthesis, sulfonated catalys:

and tri-tert-butylphonols, in the presence of ac ac-